## ADENYLYLDIDEOXYKANAMYCIN B, A PRODUCT OF THE INACTIVATION OF DIDEOXYKANAMYCIN B BY ESCHERICHIA COLI CARRYING R FACTOR

Sir :

In previous papers<sup>1~11</sup>, it was reported that acetylation, phosphorylation and adenylylation were the mechanisms of inactivaof aminoglycosidic antibiotics by tion Escherichia coli carrying R factor and Pseudomonas aeruginosa. It was found that kanamycin was inactivated by phosphorylation of the 3-hydroxyl group of 6-amino-6-deoxy-D-glucose moiety by E. coli carrying R factor, strain ML-1629<sup>3~5)</sup>, and P. aeruginosa<sup>7,10,11)</sup>. Recently, UMEZAWA et al.<sup>12)</sup> synthesized 3',4'-dideoxykanamycin B (DKB) which was effective against kanamycin-resistant E. coli ML-1629 and P. However, DKB at a aeruginosa strains. concentration of 50 µg/ml did not inhibit E. coli JR66/W677 which had been studied by BENVENISTE and DAVIES<sup>13)</sup>. They reported that this strain was obtained by transmission of R factor to sensitive strain E. coli W677, from Klebsiella pneumoniae strain type 22 # 3038 which was clinically isolated and was resistant to kanamycin, streptomycin and seven other drugs. E. coli JR66/W677 was resistant to kanamycin A (>100 µg/ml), kanamycin B (>100 µg/ml), streptomycin (50  $\mu g/ml$ ) and gentamicin (25  $\mu g/ml$ ).

In this paper, inactivation of DKB by an enzyme preparation from  $E. \ coli \ JR 66/W 677$ , and isolation and characterization of the inactivated product are presented.

The enzyme solution was prepared from logarithmically growing cells in the nutrient broth containing 5  $\mu$ g/ml of DKB. The method of OKANISHI *et al.*<sup>4)</sup> was modified as follows; growing cells of *E. coli* JR66/W677 were washed twice with 20 mM potassium phosphate buffer (pH 7.8) containing 10 mM magnesium acetate and 6 mM 2-mercaptoethanol, and suspended in an equal volume of the same buffer; the cell suspension was passed through a French pressure cell (1,200 kg/cm<sup>2</sup>) and then centrifuged at 100,000 g

for 90 minutes at 4°C; protein content of the supernatant was determined by the FOLIN method, and an enzyme solution containing 10 mg/ml of protein was prepared by dilution with the buffer described above. The optimum pH of the enzymatic reaction was 7.5 $\sim$ 8.0. In an experiment, 90 mg (0.2 mmoles) of DKB was inactivated in 500 ml of the reaction mixture containing 200 ml of the enzyme solution (2,000 mg of protein), 9,090 mg (15.1 mmoles) of disodium ATP, 800 mg (9.5 mmoles) of NaHCO<sub>3</sub>, 8,012 mg (46 mmoles) of  $K_2$ HPO<sub>4</sub>, 544 mg (4 mmoles) of KH<sub>2</sub>PO<sub>4</sub>, 1,072 mg (5 mmoles) of Mg(CH<sub>3</sub>-COO)<sub>2</sub>·4H<sub>2</sub>O and 232.3 mg (3 mmoles) of 2mercaptoethanol. After 15-hour incubation at 37°C, 100 % inactivation of DKB was confirmed by the assay of the remaining The disc-plate method using activity. Bacillus subtilis PCI 219 as a test organism was employed. After reaction, the mixture was diluted to 1,000 ml with distilled water and kept in a boiling water bath for 10 minutes. After cooling, it was filtered and the filtrate was passed through a column of Amberlite CG-50 (NH<sub>4</sub><sup>+</sup> form, 100 ml). The column was washed with 2,000 ml of water and the inactivated DKB (IDKB) on the column was eluted with 0.2 % ammonia. The eluate with positive ninhydrin reaction and UV absorption at  $254 \text{ m}\mu$  was concentrated to dryness, yielding 115 mg of a yellowish powder. The powder was rechromatographed on Amberlite CG-50, yielding 88.0 mg of slightly yellowish powder. Eighty mg of the powder was dissolved in 80 ml of 0.1 M ammonium formate and adsorbed on a column of CM-Sephadex C-25 (40 ml) equillibrated with 0.1 M ammonium formate. The column was eluted stepwise with 0.1 M, 0.2 M, 0.4 M, 0.6 M and 0.8 M ammonium formate, recording the UV absorption at 260 m $\mu$ . The fractions eluted at 0.8 M showed UV absorption. These fractions were collected and diluted with water to 2,000 ml. The diluted solution was passed through a column of Amberlite CG-50  $(NH_4^+$  form, 20 ml) and IDKB was eluted with 0.2 % ammonia. Thus, 66 mg of purified IDKB was obtained as a white powder.

IDKB darkens at 205~209°C but does not

melt at 280°C. Anal. calcd. for C<sub>28</sub>H<sub>49</sub>N<sub>10</sub>  $O_{14}P \cdot 3H_2O$ : C 40.25, H 6.64, N 16.77, P 3.71. Found: C 40.40, H 6.76, N 16.42, P 3.96. It gives positive ninhydrin, Rydon-SMITH<sup>14)</sup> and HANES<sup>15)</sup> reactions. On highvoltage paper electrophoresis at 3,500 volts for 15 minutes, using acetic acid-formic acid-water (75:25:900), IDKB moves to the cathode 13.4 cm, while DKB moves 16.8 cm. IDKB shows a UV maximum at 260  $m\mu$  ( $\varepsilon = 15,400$ ) in water, and a maximum at 258 m $\mu$  ( $\varepsilon = 14,400$ ) in 0.1 N hydrochloric IDKB is hydrolyzed to DKB and acid. adenylic acid by snake venom phosphodiesterase purchased from Boehringer Mannheim Co. Thus, IDBK is a monoadenylyl DKB. As reported in another paper<sup>16)</sup> the structure of IDKB was determined by n.m.r. analysis to be 3',4'-dideoxykanamycin B-2''adenylate, that is, the 2-hydroxyl group of the 3-amino-3-deoxy-D-glucose moiety is adenylylated.

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